

## Programme Outcomes (PO), Programme Specific Outcomes (PSO) and Course OutComes (CO)

## Department of Chemistry



Govt. College Jhandutta Distt. Bilaspur (H.P.)

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## Programme Outcomes (PO), Programme Specific Outcomes (PSO) and Course Outcomes (CO) for <u>B. Sc. With Chemistry</u>

Department of	After successful completion of three-year degree program in		
Chemistry	Chemistry a student should be able to;		
Programme	PO-1: Demonstrate various chemical phenomenons occurring in our day to		
Outcomes	day life with logical and critical thinking.		
(PO) B.Sc. (Chemistry)	<ul> <li>PO-2: Explain comprehensive knowledge and understanding of both theoretical and applied chemistry knowledge in various fields of interest like Analytical Chemistry, Physical Chemistry, Inorganic Chemistry, Organic Chemistry, Material Chemistry, etc.</li> <li>PO-3: Demonstrate Broad and balance knowledge in chemistry along with understanding of key chemical concepts, principles and theories.</li> <li>PO-4: Undertake further higher studies in chemistry and related areas or</li> </ul>		
	multidisciplinary areas that can be helpful for self- employment/entrepreneurship.		
Programme Specific	PSO-1: Develop skill and ability to got expertise in solving both		
Outcomes	theoretical and applied chemistry problems.		
(PSO)	PSO-2: To know about wide applications of chemistry in Pharma		
<b>B.Sc.</b> (Chemistry)	industries, research labs, chemical plants etc.		
	PSO-3: Learn about basic experiments of chemistry lab like		
	titration, analysis and other techniques like chromatography,		
	colorimeter, pH meter etc.		
	PSO-4: Gain the knowledge of various spectroscopic techniques		
	i.e. UV-visible, FTIR, NMR ( $^{1}$ H and $^{13}$ C) and their importance in		
	structure determination of unknown compound.		
	PSO-5: Go for higher study in Chemistry through JAM (Joint		
admission Test for M.Sc.) through various IITs & NITs and various			
State/Central universities.			
	COURSE OUTCOMES (CO): CHEMISTRY		
CLASS: B.Sc. I, II, III			
Course	e Course		
	outcomes		

CHEM101TH: ATOMIC STRUCTURE,	After studying the course, students will able to:
BONDING, GENERAL ORGANIC CHEMISTRY	1. To learn about atomic theory and its evolution.
& ALIPHATIC HYDROCARBONS	2. To learn scientific theory of atoms, concept of wave
	function.
	3. To predict the atomic structure, chemical bonding.
	and molecular geometry based on accepted models
	4 To learnhonding between atoms molecules
	interaction and energetics
	5 Explain hybridization and shapes of atomic
	5. Explain hybridization and shapes of atomic,
	distances and energies Differentiation between
	chances and energies, Differentiation between
	snape and geometry of molecules.
	6. Explain stereochemistry of organic molecules –
	conformation and configuration, asymmetric
	molecules and nomenclature.
	7. Understand concept of aromaticity, mechanism of
	aromatic reactions.
	8. Understanding hybridization and geometry of
	atoms, identifying chiral centers, enantiomers and
	diastereomers
	9. Define electrophile, nucleophiles, free radicals,
	electronegativity, resonance, hyperconjugation and
	intermediates along the reaction pathways.
CHEM102TH: STATES OF MATTER,	After studying the course, students will familiarization with:
CHEMICAL KINETICS & FUNCTIONAL	1. Various states of matter and physical properties of
ORGANIC CHEMISTRY	each stateof matter and laws related to describe the
	states.
	2. Kinetic model of gas and its properties.
	3. Maxwell distribution, mean-free path, kinetic
	energies.
	4. Behavior of real gases, its deviation from ideal
	behavior, equation of state, isotherm, and law of
	corresponding states.
	5. Liquid state and its physical properties related to
	temperature and pressure variation.
	6. Familiarization about classes of organic compounds
	and their methods of preparation.
	7. Basic uses of reaction mechanisms
	8. Preparation and uses of various classes of organic
	compounds.
	9. Organic chemistry reactions and reaction
	mechanisms.
CHEM201TH: SOLUTIONS, PHASE	After studying the course, students will familiarization with:
EQUILIBRIUM, CONDUCTANCE,	1. Understanding phases, components, Gibb's phase
ELECTROCHEMISTRY & ORGANIC	rule and itsapplications.
CHEMISTRY	2. Construction of phase diagram of different
	systems and theapplication of phase diagram.
	3. Electrolytes and electrolytic dissociation, salt
	hydrolysis andacid-base equilibria
	4. Ionic equilibria – electrolyte, ionization, dissociation.
	5. Salt hydrolysis (acid-base hydrolysis) and its
	application inchemistry.

CHEM202TH: CHEMISTRY OF MAIN GROUP	The student will able to understand and apply:
ELEMENTS. CHEMICAL ENERGETICS AND	1. Chemistry of s and p-block elements & their
EOUILIBRIA	nronerties
- (	2 Chemistry of a chic cases & their properties
	2. Chemistry of noble gases & their properties
	3. Inorganic polymers and their uses.
	4. Understanding redox reactions in hydrometallurgy
	processes.
	5. Structure, bonding of s and p block materials and
	theiroxides/compounds.
	6. Understanding chemistry of boron compounds and
	their structures.
	7. Chemistry of noble gases and their compounds;
	application of VSEPR theory in explaining structure
CHEM202TH: DASIC ANALYTICAL	and bonding.
CHEMIZUSIN: DASIC ANALITICAL CHEMISTRY	able to:
	1. Understand the basics of analytical chemistry
	2. Undertake Soil analysis, water analysis and various
	aspects food analysis
	3 Undertake analysis using chromatographic
	techniques
	4. Undertake the analysis of cosmetics like deodorants
	and antiperspirant, talcum powder etc
CHEM204TH:	After studying the course, students will familiarization with:
FUEL CHEMISTRY &	1. Energy sources (renewable and non-renewable)
<b>CHEMISTRY OF COSMETICS &amp; PERFUMES</b>	2. Coal in various industries, composition and
	processing
	3. Petroleum and Petrochemical Industry
	4. Classification of lubricants, lubricating oils,
	Properties of lubricants
	5. preparation and uses of the following: Hair dye, hair
	spray, shampoo, suntan lotions etc.
	6. Essential oils and their importance in cosmetic
	industries
CHEM301TH:	After studying the course, students will familiarization with:
POLYNUCLEAR HYDROCARBONS, DYES,	1. Polynuclear hydrocarbons, aromatic character and
HETEROCYCLIC COMPOUNDS AND	their reactions.
	2. Heterocyclic compounds and their reactions.
	3. Understanding the structure and their mechanism of
	reactions of selected polynuclear hydrocarbons.
	4. Understanding the structure, mechanism of
	reactions of selected heterocyclic compounds
	5. To study UV, IR and NMR spectroscopy.
	6. Identify structure of compound by spectroscopic methods
CHEM304TH:	The students will able to understand the:
CHEMISTRY OF TRANSITION AND INNER	1. Coordination compounds: nomenclature, theories, d-
TRANSITION ELEMENTS, COORDINATION	orbital splitting in complexes. Jhan Teller distortion.
CHEMISTRY,	chelation and stability of chelate complexes.
ORGANOMETTALICS, ACIDS & BASES	2. Transition metals, its stability, colour, oxidation
	states and stability of Transition metal complexes.
	3 Lanthanides and Actinides separation colour
	spectra and magnetic behavior lanthanide and
	actinide contraction
	4. Understanding the nomenclature of coordination
	compounds/complexes Molecular orbital theory d-
	orbital splitting in tetrahedral octahedral square
	nlanar complexes chelate effects
	5 Understanding the transition metals stability in
	reactions origin of colour and magnetic properties
	6. Know the shapes of d-orbital and degeneracy of d-
	orbital

CHEM307TH:	On completion of this course, the students will be able to:
CHEMICAL TECHNOLOGY & SOCIETY	1. Various processes and instruments used in chemical
AND BUSINESS SKILLS FOR CHEMISTRY	technology such as distillation, solvent extraction,
	extruders, pumps, mills, emulgators.
	2. Scaling up operations in chemical industry.
	3. Introduction to clean technology.
	4. Exploration of societal and technological issues
	from a chemical perspective
	5. Key business concepts: Business plans, market need
	6. Current challenges and opportunities for the
	chemistry-using industries
	7. Concept of intellectual property rights and patents.
CHEM308TH:	The students will have the knowledge of:
PESTICIDE CHEMISTRY &	1. Various pesticides, insecticides, fungicides and
PHARMACEUTICAL CHEMISTRY	herbicides.
	2. Synthesis of DDT, Malathion, Parathion, Carbofuran.
	3. Uses of Drugs & Pharmaceuticals Drug discovery
	4. Synthesis of the drugs like analgesics agents,
	antipyretic agents, anti-inflammatory agents.
	5. Production of Ethyl alcohol and citric acid,
	Penicillin, Cephalosporin, Chloromycetin and
	Streptomycin, Lysine, Glutamic acid, Vitamin B2,
	Vitamin B12 and Vitamin C.